

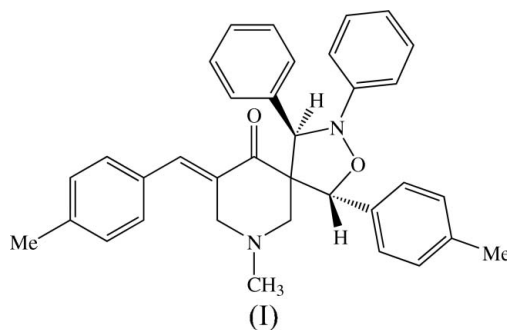
J. Suresh,^a R. Ranjith Kumar,^b
S. Perumal,^b A. Mostad^c and
S. Natarajan^{d*}^aDepartment of Physics, The Madura College,
Madurai 625 011, India, ^bSchool of Chemistry,
Madurai Kamaraj University, Madurai 625 021,
India, ^cDepartment of Chemistry, University of
Oslo, PO Box 1033, Blindern, N-0315 Oslo 3,
Norway, and ^dDepartment of Physics, Madurai
Kamaraj University, Madurai 625 021, IndiaCorrespondence e-mail:
s_natarajan50@yahoo.com

Key indicators

Single-crystal X-ray study
 $T = 105\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.043
 wR factor = 0.113
Data-to-parameter ratio = 19.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.7-Methyl-9-(4-methylbenzylidene)-1-(4-methyl-
phenyl)-3,4-diphenyl-2-oxa-3,7-diazaspiro[4.5]-
decan-10-oneIn the title compound, $\text{C}_{35}\text{H}_{34}\text{N}_2\text{O}_2$, the isoxazolidine ring
adopts a twisted conformation and the piperidine ring is in a
half-chair conformation. Weak $\text{C}-\text{H}\cdots\pi$ interactions are
observed in the crystal structure.Received 17 October 2006
Accepted 26 October 2006

Comment

The 1,3-dipolar cycloaddition of nitrones with olefinic dipolarophiles proceeds through a concerted mechanism yielding highly substituted isoxazolidines (Gothelf & Jorgensen, 1998). Isoxazolidines are potential precursors for biologically important compounds such as amino sugars (Gothelf & Jorgensen, 1998), alkaloids (Goti *et al.*, 1997), β -lactams (Ali *et al.*, 1988) and amino acids (Annuziata *et al.*, 1987), and exhibit antibacterial and antifungal activities (Kumar *et al.*, 2003). Highly substituted spiro-isoxazolidines result from the 1,3-dipolar cycloaddition of exocyclic olefins with nitrones and these spiro-isoxazolidines have also been transformed into complex heterocycles (Colombi *et al.*, 1978). Heterocycles with piperidine sub-structures display important biological activities, such as anticancer (Dimmock *et al.*, 2001) and cytotoxic (El-Subbagh *et al.*, 2000), besides being useful as synthons in the construction of alkaloid natural products (Lee *et al.*, 2001). The above importance of isoxazolidine and piperidine sub-structures led us to investigate the cycloaddition of 1-methyl-3,5-bis[(*E*)-(4-methylphenyl)methylidene]tetrahydro-4(1*H*)-pyridinone with *C,N*-diphenylnitron. The present work reports the X-ray crystallographic study of one such substituted isoxazolidine, the title compound, (I).



The molecular structure of (I) is shown in Fig. 1. The isoxazolidine ring adopts a twisted conformation with a pseudo-twofold axis passing through atom C4 and the mid-point of the C1—O2 bond. The piperidine ring adopts a half-chair conformation with atoms N7 and C6 deviating by $-0.452(2)$ and $0.346(2)$ Å, respectively, from the least-squares plane defined by C5/C8/C9/C10. The dihedral angle

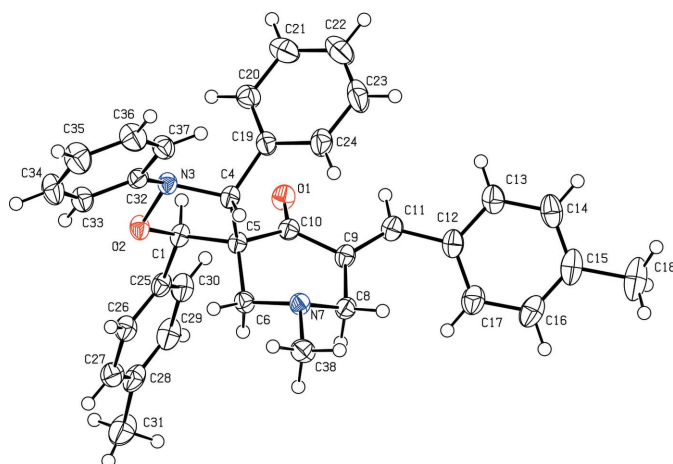


Figure 1
The molecular structure of (I), showing 50% probability displacement ellipsoids and the atomic numbering scheme.

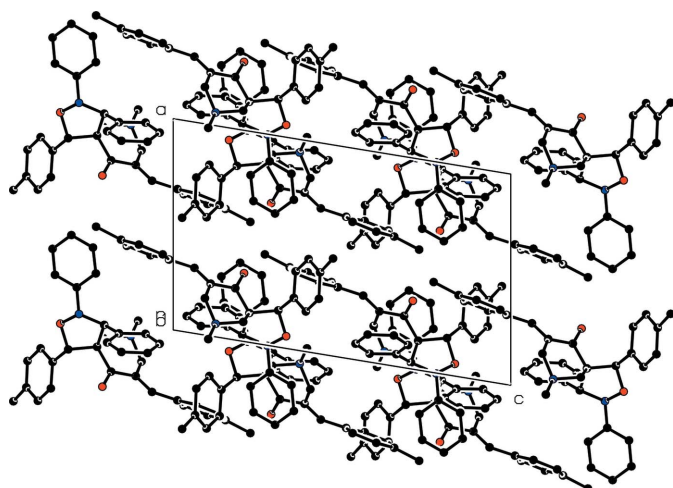


Figure 2
Packing diagram of (I). H atoms have been omitted for clarity.

between the C19–C24 and C32–C37 phenyl rings is $79.12(6)^\circ$. The orientation of the (4-methylphenyl)methylidene substituent with respect to the attached piperidine ring may be influenced by the intramolecular C11–H11 \cdots O1 interaction (Table 1). The dihedral angle between the C5/C8/C9/C10 and C12–C17 planes is $48.77(7)^\circ$.

In the crystal structure (Fig. 2), two weak C–H \cdots π interactions, *viz.* C26–H26 \cdots Cg1ⁱ and C38–H38B \cdots Cg2ⁱⁱ (Cg1 and Cg2 are the centroids of the rings C19–C24 and C12–C17, respectively; symmetry codes are given in Table 1) are observed. There are no π – π interactions or intermolecular C–H \cdots O interactions.

Experimental

A mixture of 1-methyl-3,5-bis[(*E*)-(4-methylphenyl)methylidene]-tetrahydro-4(1*H*)-pyridinone (1 mol) and nitron (4 mol) in toluene (15 mol) was refluxed for 10 h. The product was then purified by column chromatography employing petroleum ether–ethyl acetate as

eluant and subsequently recrystallized from ethanol (yield 39%, m.p. 447–448 K).

Crystal data

C₃₅H₃₄N₂O₂
M_r = 514.64
 Monoclinic, *P*2₁/*c*
a = 11.131 (2) Å
b = 14.206 (3) Å
c = 18.016 (4) Å
 β = 99.32 (3)°
V = 2811.2 (10) Å³

Z = 4
D_x = 1.216 Mg m^{−3}
 Mo *K*α radiation
 μ = 0.08 mm^{−1}
T = 105 (2) K
 Block, colourless
 0.28 × 0.14 × 0.12 mm

Data collection

Bruker SMART APEX CCD
 diffractometer
 ω scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 1998)
T_{min} = 0.987, *T_{max}* = 0.991

41834 measured reflections
 6967 independent reflections
 5288 reflections with *I* > 2σ(*I*)
R_{int} = 0.050
 θ_{\max} = 28.3°

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.043
wR (*F*²) = 0.113
S = 1.02
 6967 reflections
 355 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0448P)^2 + 1.0249P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 (Δ/σ)_{max} = 0.001
 $\Delta\rho_{\max}$ = 0.27 e Å^{−3}
 $\Delta\rho_{\min}$ = −0.22 e Å^{−3}

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> –H \cdots <i>A</i>	<i>D</i> –H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> –H \cdots <i>A</i>
C11–H11 \cdots O1	0.93	2.40	2.7712 (17)	104
C26–H26 \cdots Cg1 ⁱ	0.93	2.97	3.7137 (17)	138
C38–H38B \cdots Cg2 ⁱⁱ	0.96	2.95	3.7646 (18)	144

Symmetry codes: (i) $-x, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x, -y + 2, -z$.

H atoms were placed in calculated positions (C–H = 0.93–0.98 Å), and allowed to ride on their carrier atoms, with *U*_{iso}(H) = 1.2*U*_{eq}(C) for CH₂ and CH groups, and 1.5*U*_{eq}(C) for CH₃ groups.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97.

The authors thank the UGC for the SAP programme. JS thanks the UGC and the management of The Madura College, Madurai, for providing a teacher fellowship. SP thanks CSIR, New Delhi, for a major research project.

References

- Ali, A. S., Khan, J. H. & Wazeer, M. I. M. (1988). *Tetrahedron*, **44**, 5911–5920.
 Annuziata, R., Chinquini, M., Cozzi, F. & Raimondi, L. (1987). *Tetrahedron*, **43**, 4051–4056.
 Bruker (1998). SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (2001). SMART and SAINT. Version 6.12. Bruker AXS Inc., Madison, Wisconsin, USA.
 Colombi, S., Vecchio, G., Gottarelli, G., Samori, B., Lanfredi, A. M. M. & Tiripicchio, A. (1978). *Tetrahedron*, **34**, 2967–2976.

- Dimmock, J. R., Padmanilayam, M. P., Puthucode, R. N., Nazarali, A. J., Motaganahalli, N. L., Zello, G. A., Quail, J. W., Oloo, E. O., Kraatz, H. B., Prisciak, J. S., Allen, T. M., Santos, C. L., Balzarini, J., De Clercq, E. & Manavathu, E. K. (2001). *J. Med. Chem.* **44**, 586–593.
- El-Subbagh, H. I., Abu-Zaid, S. M., Mahran, M. A., Badria, F. A. & Al-Obaid, A. M. (2000). *J. Med. Chem.* **43**, 2915–2921.
- Gothelf, K. V. & Jorgensen, K. A. (1998). *Chem. Rev.* **98**, 863–909.
- Goti, A., Fedi, V., Nanneli, L., De Sarlo, F. & Brandi, A. (1997). *Synlett*, pp. 577–579.
- Kumar, K. R. R., Mallesha, H. & Rangappa, K. S. (2003). *Synth. Commun.* **33**, 1545–1555.
- Lee, H. K., Chun, J. S. & Pak, C. S. (2001). *Tetrahedron Lett.* **42**, 3483–3486.
- Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Gottingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.